

## METABOLIC NMR WITHOUT THE MAGNET

James Eills,<sup>a,b,c</sup> Román Picazo-Frutos,<sup>b,c</sup> Oksana Bondar,<sup>d</sup> Eleonora Cavallari,<sup>d</sup>  
 Carla Carrera,<sup>e</sup> Sylwia Barker,<sup>f</sup> Marcel Utz,<sup>f</sup> Silvio Aime,<sup>d</sup> Francesca Reineri,<sup>d</sup>  
 Dmitry Budker,<sup>b,c,g</sup> John W. Blanchard<sup>h</sup>

<sup>a</sup> Institute for Bioengineering of Catalonia, Barcelona, Spain

<sup>b</sup> Johannes-Gutenberg University, Mainz, Germany

<sup>c</sup> Helmholtz-Institute Mainz, Mainz, Germany

<sup>d</sup> Center of Molecular Imaging, University of Turin, Turin, Italy

<sup>e</sup> Institute of Biostructures and Bioimaging, Turin, Italy

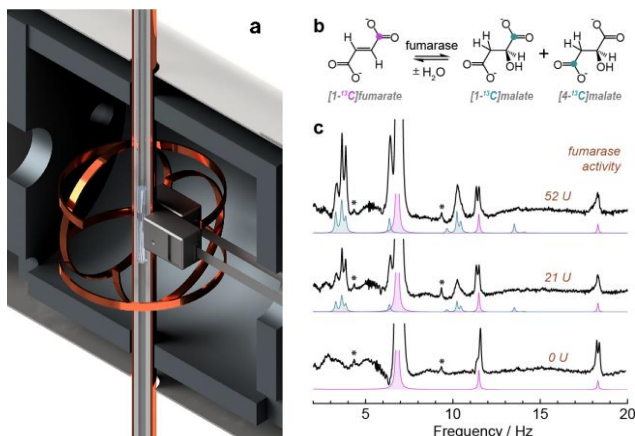
<sup>f</sup> University of Southampton, Southampton, U.K.

<sup>g</sup> University of California at Berkeley, Berkeley, U.S.A.

<sup>h</sup> NVision Imaging Technologies, Ulm, Germany

✉ jeills@ibecbarcelona.eu

Zero- to ultralow-field (ZULF) NMR is a modality of NMR experiment performed in the absence of a strong magnetic field. In this regime, Larmor precession is suppressed, and other interactions such as J-couplings dominate. This grants three important advantages: the low frequency signals readily penetrate metals and conductive materials, magnetic susceptibility-induced line broadening from sample inhomogeneity is suppressed, and no bulky superconducting magnet is needed. In this work we form the biomolecules  $[1-^{13}\text{C}]$ fumarate and  $[1-^{13}\text{C}]$ pyruvate in aqueous solution via parahydrogen-induced polarization. We acquire the low-frequency ZULF signals using optical magnetometers, with Helmholtz coils surrounding the sample to apply magnetic field pulses. This is all contained within a mu-metal shield to attenuate Earth's field. With this apparatus we are able to acquire zero-, ultralow-, and low-field spectra. The conversion of fumarate into malate, and pyruvate into lactate, are metabolic processes observed *in vivo* in hyperpolarization-enhanced magnetic resonance imaging experiments.<sup>[1,2]</sup> We show that ZULF NMR can be used to study metabolism by observing these two biochemical reactions. This work paves the way to a heretofore unexplored class of biomedical imaging applications.



**Figure 1.** (a) The ZULF setup used in this work. (b) The transformation of fumarate into malate. (c) Zero-field spectra of hyperpolarized  $[1-^{13}\text{C}]$ fumarate after addition of the enzyme fumarase (given in enzyme units U), with simulated fumarate and malate spectra beneath.

**Acknowledgements:** This work has been supported by European Union's Horizon 2020 Research and Innovation Programme.

### REFERENCES

- [1] S. J. Nelson et al., *Sci. Transl. Med.* **2013**, 5(198), 198ra108.
- [2] M. R. Clatworthy, M. I. Kettunen, D.-E. Hu, R. J. Mathews, T. H. Witney, B. W. C. Kennedy, S. E. Bohndiek, F. A. Gallagher, L. B. Jarvis, K. G. C. Smith, K. M. Brindle, *Proc. Natl. Acad. Sci.*, **2012**, 109(33), 13374–13379.